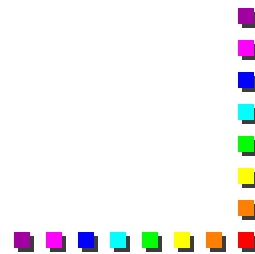


Cryostat Options and Heat Loads

Astronomy 525

Lecture 25



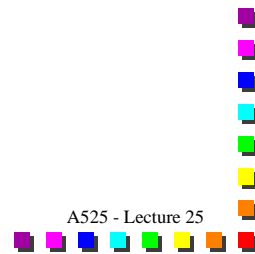
Outline

- Temperature Requirements
- Cooling System Options
- Cryogenic Capacities
- Cryostat examples
- Heat Loads
 - Radiation
 - Shields
 - Window

Heat Loads

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Temperature Requirements

- Bolometers: minimize heat capacity and thermal conductance \Rightarrow smaller the better
- Photodetective devices: Dark current \ll photocurrent
 - The generation of dark current is an excitation process with

$$i_d \propto \exp\left(-\frac{\Delta E}{kT}\right)$$

Some rough operating temperatures for a selected sample of detectors are given on the right.

Material	$\lambda_c(\mu\text{m})$	T(K)
Si CCD	1.0	120
HgCdTe	2.5	< 65
InSb	5.0	< 35
Si:As	26.	< 8
Ge:Ga	120	< 2

Heat Loads

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Temperature Reqms (cont'd)

- Dewar photon power, or “background” \ll signal power, or photocurrent
 - The radiation from the structures and optics surrounding the detector should be much less than the signal.

$$N_B (\text{ph/sec}) = \int_{\nu_{\min}}^{\nu_{\max}} \frac{\pi B_\nu(T) A_d}{h\nu} d\nu \ll N_S$$

N_S = signal photon rate on detector

A_d = detector area

ν_{\min}, ν_{\max} = limits of detector response

Heat Loads

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Cooling System Considerations

- Cooling capacity
- Operating temperature
- Desired “hold-time” (lifetime)
- Weight, volume, and power
- Device dimensions
- Cool down time
- Operating environment

Heat Loads

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Cooling System Options

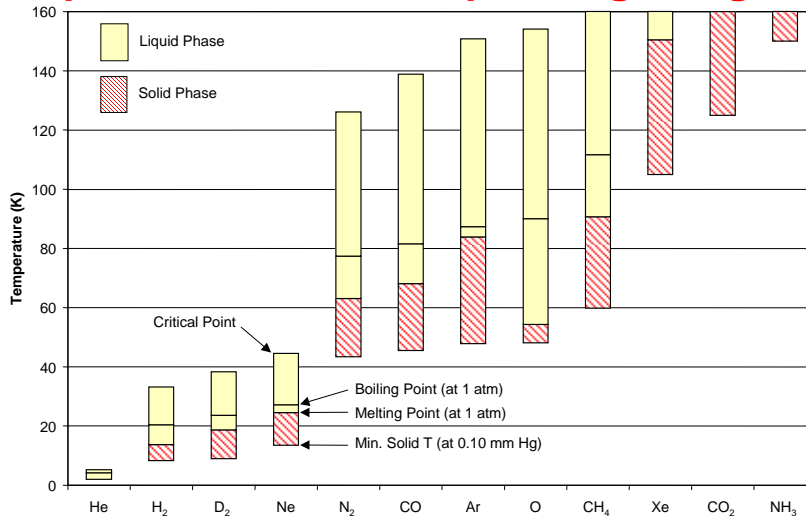
	Range (K)	Capacity (W)
■ Open-cycle		
■ Liquid or solid cryogenes	4.2 - 200	Unlimited
■ High pressure gas expansion	4.2 - 90	< 20
■ Closed-cycle system		
■ Mechanical refrigerators	4 - 77	< 15
■ ³ He refrigerators	0.22 - 0.3	< 0.001
■ ADR		
■ Adiabatic Demag. Refrigerators	0.05	<0.000001
■ Thermoelectric coolers		
■ Single stage	230 - 300	0.1 - 100
■ Multi-stage	145 - 230	0.01 - 1.0
■ Passive radiators		
■ Space environments	35 - 200	1 to 10

Heat Loads

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Expendable Coolant Operating Ranges



Heat Loads

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Cryogen Properties

Cryo- gen	Melting Point at 0.1mm Hg	Melting Point at 1 atm	Boiling Point at 1 atm	Critical Temp.	Critical Pressure	Liquid Density at bp	Vapor Pressure Solid at mp	Heat of Vapor at bp	Heat of Fusion at mp	Density of Solid	12 month holdtime for liquid with 100 mW load	
	(K)	(K)	(K)	(K)	(atm)	(kg/l)	(mm Hg)	(J/g)	(J/g)	(kg/l)	Mass(kg)	Vol. (l)
He	2.0	2.0	4.2	5.2	2.3	0.125	--	20.5	4.18	0.129	153.94	1231.52
H ₂	8.3	13.7	20.4	33.2	13.0	0.071	54	448.0	58.15	0.087	7.04	99.21
D ₂	9.0	18.7	23.6	38.3	16.2	0.173	12.8	286.0	50.00		11.03	63.78
Ne	13.5	24.5	27.2	44.5	26.8	1.200	323	87.0	16.72	1.490	36.27	30.23
N ₂	43.4	63.1	77.4	126.1	33.5	0.808	96.5	199.0	25.52	1.020	15.86	19.63
CO	45.5	68.1	81.6	138.8	35.0	0.812	--	213.0	29.27	1.030	14.82	18.25
Ar	47.8	83.9	87.4	150.8	48.0	1.391	516	162.7	28.05	1.670	19.40	13.94
O ₂	48.1	54.3	90.1	154.1	50.1	1.140	2	212.5	13.80	1.359	14.85	13.03
CH ₄	59.8	90.7	111.7	190.5	45.8	0.425	71	581.0	60.25	0.520	5.43	12.78
Xe	105.0	150.5	165.3	290.0	58.0	3.100	615	96.3	119.10	3.540	32.79	10.58
CO ₂	125.0	217.5	194.6	304.5	73.0	1.510	--	574.0	71.60	1.700	5.50	3.64
NH ₃	150.0	195.4	239.8	405.0	111.2	0.683	45	1363.0	--	0.800	2.32	3.39

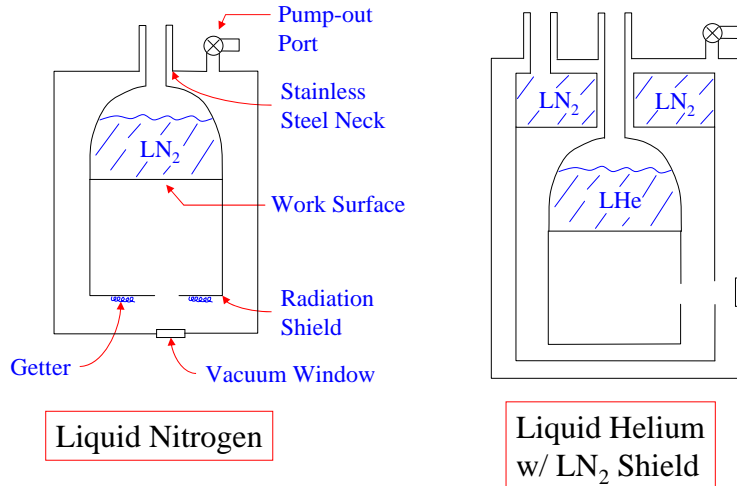
See Infrared & Electro-Optical Handbook, Volume 3, Chapter 6 for more details on cryogen properties, mechanical coolers, thermoelectric coolers, and radiative cooling.

Heat Loads

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Basic Liquid Cryogen Cryostats



Heat Loads

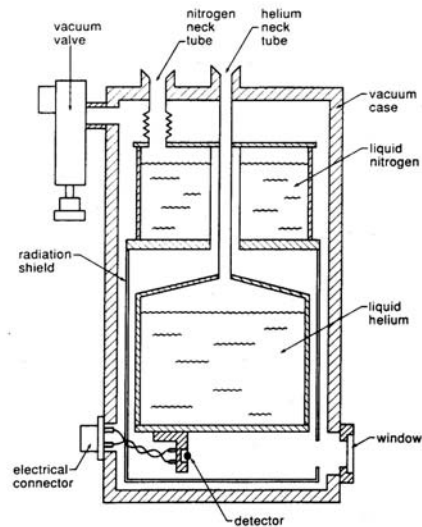
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Low Temperature Cryostats

Simple Helium-4 dewar

- $T_{\text{ambient Pressure}} = 4.2 \text{ K}$
- $T_{\text{pumped}} = 1 \text{ to } 2 \text{ K}$
- Easy to 1.3 K with vapor pressure $\sim 3 \text{ mm Hg}$
- More difficult to 1.1 K with vapor pressure $\sim 0.5 \text{ mm Hg}$ due to superfluid ^4He that may creep up the neck, causing a thermal short due to its very high thermal conductivity



Heat Loads

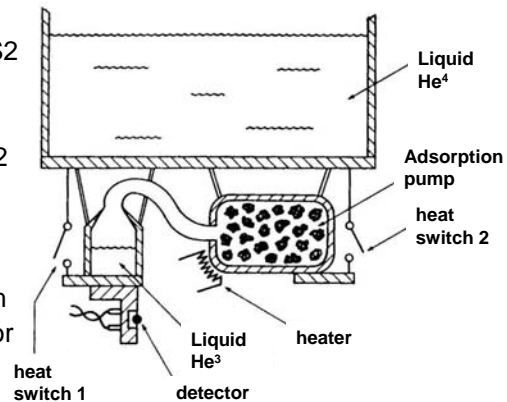
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Low Temperature Cryostats

Helium-3 closed cycle refrigerator

- Heat switch 1 closed, HS2 open, heater on:
⇒ condensed ^3He
- Heat switch 1 open, HS2 closed, heater off:
⇒ pumps ^3He
⇒ $T \rightarrow 0.3\text{ K}$
- The pumping mechanism is the affinity of carbon for ^3He



Heat Loads

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Dual Stage ^3He Refrigerator

- First stage acts as a thermal guard for the inner stage
- Temperatures as low as 220 mK attainable
- This model is used in ZEUS at Cornell
- Manufactured by Chase Instruments



Heat Loads

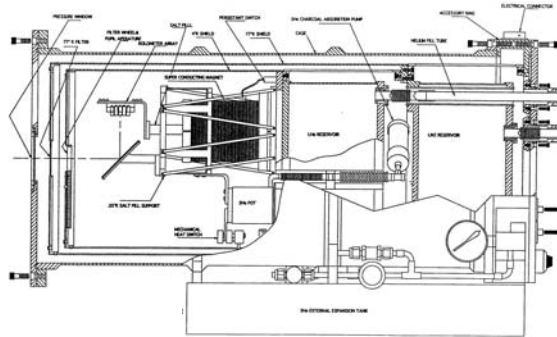
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Low Temperature Cryostats

Adiabatic demagnetization refrigerator (ADR)

- Multistage system
- Salt pill holds > 40 hours
- Used in SPIFI



- Paramagnetic salt (ferric ammonium alum) cooled to 1.5 K with ⁴He cryostat
- ³He condenses, superconducting magnet energized to align magnetic domains, pill cooled again to 1.5 K (B fields are around 4 Tesla!)
- Heat switch between ⁴He 1.5 K reservoir and ³He pot and salt pill opened
- Current in magnet dissipated ⇒ salt pill domains randomize ⇒ T → 50 mK

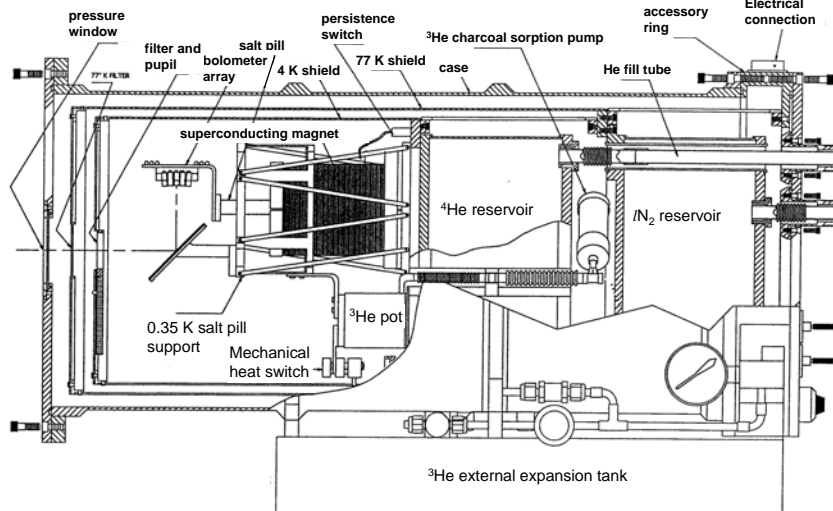
Heat Loads

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Low Temperature Cryostats

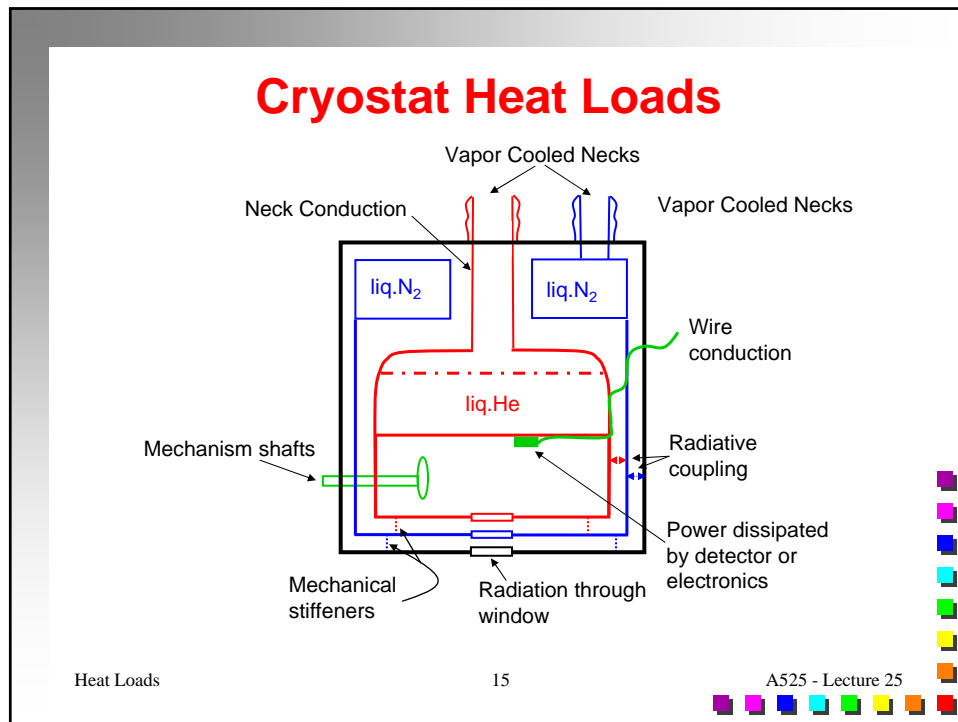


Heat Loads

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Cryogen Boil-off Rates

Cryogen	T (K)	Cryogen per Watt-Hour		Power for 1 liter/day (mW)
		Mass (g)	Volume (liters)	
LHe	4	176	1.4	30
LH2	20	8	0.11	380
LN2	77	18	0.022	1900
SH2	7-10	7	0.079	530

- Per liter, liquid nitrogen has a very good cooling capacity – good for ground based systems
- Per gram, liquid and solid hydrogen are the best! – good for space based applications

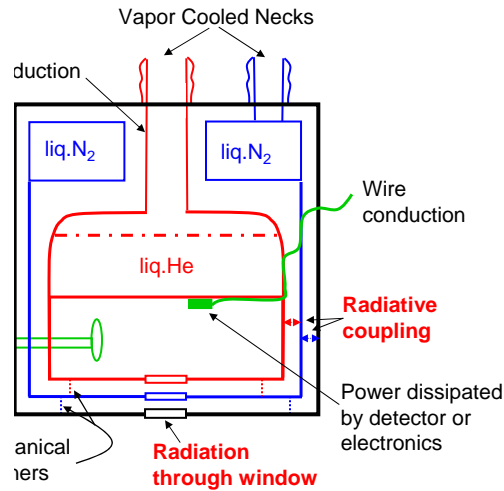
Heat Loads

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Radiative Energy Transfer

- The primary radiative loads into the cryostat are
 - Radiative coupling between the walls/shields for each stage of the cryostat.
 - Photons entering through the dewar window.



Heat Loads

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Shield Coupling

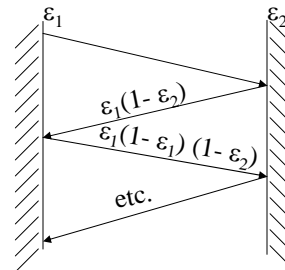
- For two parallel surfaces the net flux transferred from surface #1 to #2 is proportional to

$$R = \varepsilon_1 \varepsilon_2 \left[1 + (1 - \varepsilon_1)(1 - \varepsilon_2) + (1 - \varepsilon_1)^2 (1 - \varepsilon_2)^2 + \dots \right]$$

So that:

$$R = \frac{\varepsilon_1 \varepsilon_2}{1 - (1 - \varepsilon_1)(1 - \varepsilon_2)}$$

$$= \frac{1}{\frac{1}{\varepsilon_1} + \frac{1}{\varepsilon_2} - 1}$$



The appropriate factor for surface #2 to #1 must be the same by conservation of energy

Heat Loads

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Shield Coupling (cont'd)

- The net transfer of energy between the two surfaces of area A is

$$Q_{rad} = \frac{A}{\frac{1}{\epsilon_1} + \frac{1}{\epsilon_2} - 1} \sigma (T_1^4 - T_2^4)$$

- Clearly Q_{rad} is smallest when $\epsilon_1, \epsilon_2 \ll 1$.
- How large is the radiative coupling? Taking $T_1 = 300$ K, $T_2 = 4.2$ K, and with $\sigma = 5.67 \times 10^{-8}$ W/m²/K⁴ gives:

$$Q_{rad} = \frac{459A}{\frac{1}{\epsilon_1} + \frac{1}{\epsilon_2} - 1} \text{ W/m}^2 \longrightarrow 230A\epsilon \text{ W/m}^2$$

$\epsilon_1, \epsilon_2 = \epsilon \ll 1$

Heat Loads

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Shield Coupling (cont'd)

- For a small dewar the surface area might be ~ 0.2 m². Then with $\epsilon_1 = \epsilon_2 = \epsilon \ll 1$

$$Q_{rad} = 46\epsilon \text{ Watts}$$

- For $Q_{rad} \leq 30$ mW (one liter of LHe/day), $\epsilon \leq 6.5 \times 10^{-4}$
 - This is extremely difficult, which is why a colder radiation shield is used around the LHe reservoir.
- If the coupling is between 300 K and 78 K (LN₂), the equation for Q_{rad} is essentially the same.
 - For $Q_{rad} \leq 1.9$ W (one liter of LN₂/day), $\epsilon \leq 0.042$, which is much more reasonable.

Heat Loads

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Shield Coupling (cont'd)

- The radiative coupling between the LN₂ and LHe cans is

$$Q_{rad} = 0.21 \varepsilon \text{ Watts}$$

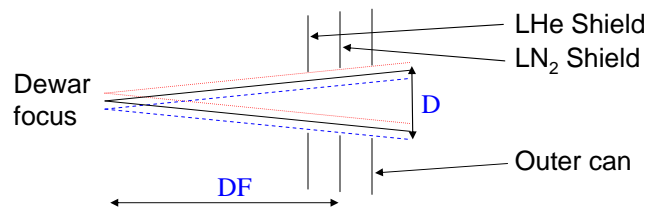
- For $Q_{rad} \leq 30 \text{ mW}$ (one liter of LHe/day), $\varepsilon \leq 0.14$
- This contribution to the heat input to the LHe reservoir can be made small.

Heat Loads

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Window Heat Load



- A typical design might have $DF \sim 15 \text{ cm}$. For an $f/12$ beam with a 10 arcsec FOV at a 10.4-m telescope, the size of the window is:

$$D \sim \frac{15}{12} + \frac{10}{57.3 \cdot 60 \cdot 60} \cdot 1040 \cdot 12 = 1.86 \text{ cm}$$

$$P \approx A \sigma T^4 (\Omega/\pi) \sim 36 \text{ mW} \quad \Omega \sim 1 \text{ sr}$$

- This can be important – especially if the power is deposited on the helium surface
- One can use blocking filters that reflect the power back out the window

Heat Loads

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